

**Registration form**

**WATER TREATMENT 8 HOURS CEU COURSE  
48 HOUR RUSH ORDER PROCESSING FEE ADDITIONAL \$50.00**

**Start and Finish Dates:** \_\_\_\_\_

*You will have 90 days from this date in order to complete this course*

List number of hours worked on assignment must match State Requirement. \_\_\_\_\_

**Name** \_\_\_\_\_ **Signature** \_\_\_\_\_

*I have read and understood the disclaimer notice on page 2. Digitally sign XXX*

**Address** \_\_\_\_\_

**City** \_\_\_\_\_ **State** \_\_\_\_\_ **Zip** \_\_\_\_\_

**Email** \_\_\_\_\_ **Fax (\_\_\_\_\_)** \_\_\_\_\_

**Phone:**  
**Home (\_\_\_\_\_)** \_\_\_\_\_ **Work (\_\_\_\_\_)** \_\_\_\_\_

**Operator ID #** \_\_\_\_\_ **Exp. Date** \_\_\_\_\_

**Class/Grade** \_\_\_\_\_

**Please circle/check which certification you are applying the course CEU's.**

Water Treatment \_\_\_ Water Distribution \_\_\_ Other \_\_\_\_\_

**Technical Learning College TLC PO Box 3060, Chino Valley, AZ 86323  
Toll Free (866) 557-1746 Fax (928) 272-0747 [info@tlch2o.com](mailto:info@tlch2o.com)**

If you've paid on the Internet, please write your Customer# \_\_\_\_\_

Please invoice me, my PO# \_\_\_\_\_

***We will stop mailing the certificate of completion, so we need either your fax number or e-mail address. We will e-mail the certificate to you, if no e-mail address; we will fax it to you.***

# DISCLAIMER NOTICE AND AFFIDAVIT OF EXAM COMPLETION

## Technical Learning College CEU Course Policies

I understand that it is my responsibility to ensure that this Continuing Education Unit (CEU) course is either approved or accepted in my State for CEU credit. I acknowledge that State laws and rules change frequently, and I believe this course is currently accepted in my State for CEU or contact hour credit. If it is not, I will not hold Technical Learning College responsible. Furthermore, I recognize that this type of study program deals with dangerous conditions, and I will not hold Technical Learning College, Technical Learning Consultants, Inc. (TLC), The Love of Christ Bible College liable for any errors, omissions, or advice contained in this CEU education training course, nor for any violation, injury, neglect, or damage caused by this CEU education training, course material suggestion, or error. If I require assistance, I will contact TLC and double-check that my registration page and assignment have been received and graded.

## State Approval Information

Please refer to the State Approval Listing Link to verify if your State accepts or has pre-approved this course. Note that not all States or courses are listed. If the course is not accepted for CEU credit, Technical Learning College will provide the course free of charge if you request your State to accept it for credit.

## State Approval Listing URL:

<http://www.tlch2o.com/downloads/PDF/CEU%20State%20Approvals.pdf>

Course Manual-Printed Version

You may obtain a printed version of the course manual from TLC for an additional \$129.95, plus shipping charges.

## Affidavit of Exam Completion

I affirm that I personally completed the entire text of the course. I also affirm that I completed the exam without assistance from any outside source. I understand that it is my responsibility to file or maintain my certificate of completion as required by the State or by the designation organization.

## Grading Information

To maintain the integrity of our courses, we do not distribute test scores, percentages, or information on questions missed. Our exams are based on pass/fail criteria, with the benchmark for successful completion set at 70%. Once you pass the exam, your record will reflect a successful completion, and a certificate will be issued to you.

## Rush Grading Service

If you need your assignment graded and results mailed to you within a 48-hour period, an additional rush service handling fee of \$50.00 will apply. Please note that this fee may not cover postage costs. To request this service, write "RUSH" at the top of your Registration Form. We will prioritize your assignment in the grading and processing line.

For security purposes, please fax or e-mail a copy of your driver's license and always call us to confirm we've received your assignment and to confirm your identity.

## CERTIFICATION OF COURSE PROCTOR

If required by the State, Technical Learning College requires that students who takes a correspondence or home study program course must pass a proctored course reading, quizzes and final examination. The proctor must complete and provide the school with a certification form approved by the commission for each examination administered by the proctor.

**Instructions.** When a student completes the course work, fill out the blanks in this section and provide the form to the proctor with the examination.

Name of Course: \_\_\_\_\_

Name of Licensee: \_\_\_\_\_

**Instructions to Proctor.** After an examination is administered, complete and return this certification and examination to the school in a sealed exam packet or in pdf format.

I certify that:

1. I am a disinterested third party in the administration of this examination. I am not related by blood, marriage or any other relationship to the licensee which would influence me from properly administering the examination.
2. The licensee showed me positive photo identification prior to completing the examination.
3. The enclosed examination was administered under my supervision on \_\_\_\_\_. The licensee received no assistance and had no access to books, notes or reference material.
4. I have not permitted the examination to be compromised, copied, or recorded in any way or by any method.
5. Provide an estimate of the amount of time the student took to complete the assignment.

Time to complete the entire course and final exam. \_\_\_\_\_

Notation of any problem or concerns:

Name and Telephone of Proctor (please print):

\_\_\_\_\_

\_\_\_\_\_

Signature of Proctor



# Water Treatment 8 Hours Course Answer Key

Name \_\_\_\_\_

Phone \_\_\_\_\_

Did you check with your State agency to ensure this course is accepted for credit?

You are responsible to ensure this course is accepted for credit. No refunds.  
Method of Course acceptance confirmation. Please fill this section

Website \_\_ Telephone Call\_\_ Email\_\_\_\_ Spoke to\_\_\_\_\_

Did you receive the approval number, if applicable? \_\_\_\_\_

What is the course approval number, if applicable? \_\_\_\_\_

*You can electronically complete this assignment in Adobe Acrobat DC.*

Please Circle, Bold, Underline or X, one answer per question. A **felt tipped pen** works best.

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*Please e-mail or fax this survey along with your final exam*

**WATER TREATMENT 8 HOURS CEU COURSE  
CUSTOMER SERVICE RESPONSE CARD**

NAME: \_\_\_\_\_

E-MAIL \_\_\_\_\_ PHONE \_\_\_\_\_

***PLEASE COMPLETE THIS FORM BY CIRCLING THE NUMBER OF THE  
APPROPRIATE ANSWER IN THE AREA BELOW.***

Please rate the difficulty of your course.

Very Easy 0 1 2 3 4 5 Very Difficult

Please rate the difficulty of the testing process.

Very Easy 0 1 2 3 4 5 Very Difficult

Please rate the subject matter on the exam to your actual field or work.

Very Similar 0 1 2 3 4 5 Very Different

How did you hear about this Course? \_\_\_\_\_

What would you do to improve the Course?

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Any other concerns or comments.

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***Please write down any questions you were not able to find the answers to or that  
have errors.***

## Disclaimer Notice

**Amount of Time for Course Completion – How many hours you spent on course?**

**Must match State Hour Requirement \_\_\_\_\_ (Hours)**

*I understand that I am 100 percent responsible for ensuring that TLC receives the Assignment and Registration Key. I understand that TLC has a zero tolerance towards not following their rules, cheating or hostility towards staff or instructors. I need to complete the entire assignment for credit. There is no credit for partial assignment completion. My exam was proctored.*

*I will contact TLC if I do not hear back from them within 2 days of assignment submission. I will not hold TLC liable for any errors, injury, death or non-compliance with rules. I will abide with all federal and state rules and rules found on page 2. I will forfeit my purchase costs and will not receive credit or a refund if I do not abide by TLC's rules.*

**Please Sign that you understand and will abide with TLC's Rules.**

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Signature

## When Finished with Your Assignment...

### REQUIRED DOCUMENTS

Please scan the **Registration Page, Answer Key, Proctoring report, Survey and Driver's License** and email these documents to [info@TLCH2O.com](mailto:info@TLCH2O.com).

### IPhone Scanning Instructions

If you are unable to scan, take a photo of these documents with your **iPhone** and send these photos to TLC, [info@TLCH2O.com](mailto:info@TLCH2O.com).

### FAX

If you are unable to scan and email, please fax these documents to TLC, if you fax, call to confirm that we received your paperwork. **(928) 468-0675**

### Rush Grading Service

If you need this assignment graded and the results mailed to you within a 48-hour period, prepare to pay an additional rush service handling fee of \$50.00.

## Water Treatment 8 Hours CEU Training Course Assignment

The Water Treatment 8 Hours CEU course assignment is available in Word on the Internet for your convenience, please visit [www.abctlc.com](http://www.abctlc.com) and download the assignment and e-mail it back to TLC.

You will have 90 days from receipt of this manual to complete it in order to receive your Professional Development Hours (PDHs) or Continuing Education Unit (CEU). A score of 70 % or better is necessary to pass this course. If you should need any assistance, please email or fax all concerns and the completed ANSWER KEY to [info@tlch2o.com](mailto:info@tlch2o.com).

Select one answer per question. Please utilize the answer key. (s) on the answer will indicate either plural or singular tenses.

***Please write down any questions you were not able to find the answers to or that have errors.***

### Hyperlink to the Glossary and Appendix

<http://www.abctlc.com/downloads/PDF/WTGlossary.pdf>

### Three Types of Public Water Systems

1. There are approximately 85,000 systems
- A. GCWS      C. NTNCWSs  
B. CVSs      D. None of the above

### Water Quality Section

#### Surface (Raw) Water Introduction

2. Operators need to appropriately treat surface water is never pure of \_\_\_\_\_, it. Most of the earth's water sources obtain their water supplies through precipitation.

- A. Excess nutrients      C. Pollution  
B. Biological actions      D. None of the above

3. Raw water generally contains varying amounts of dissolved minerals including calcium, magnesium, sodium, chlorides, sulfates and bicarbonates, depending on its source.

- A. True      B. False

#### Surface Water Properties

4. Adjustments in the dissolved oxygen, algae, temperature, suspended solids, turbidity, and carbon dioxide will change because of \_\_\_\_\_.

- A. Excess nutrients      C. Discharge  
B. Biological activities      D. None of the above

5. Water is accepted as the \_\_\_\_\_ because will dissolve most substances that comes in contact.

- A. Universal solvent      C. Surface water  
B. Water quality      D. None of the above

## Managing Water Quality at the Source

6. Contingent upon the region, source water may have several restrictions of use as part of a Water Shed Management Plan. In some areas, it may be restricted from recreational use, discharge or runoff from agriculture, or \_\_\_\_\_.

- A. Excess nutrients
- B. Biological actions
- C. Industrial and wastewater discharge
- D. None of the above

## Physical Characteristics of Water

7. Physical characteristics are the elements found that are considered alkali, metals, and non-metals such as carbonates, fluoride, \_\_\_\_\_. The consumer relates it to scaling of faucets or staining.

- A. pH and alkalinity
- B. Sulfides or acids
- C. Powdered activated carbon and chlorine
- D. None of the above

8. Total Dissolved Solids (TDS) is not a primary pollutant; it is a gauge of appealing water characteristics such as hardness and an indication of an assortment of chemical contaminants that might be present, such as?

- A. Turbidity
- B. Colloids
- C. Arsenic
- D. None of the above

9. pH is the negative logarithm of the hydrogen ion concentration,  $[H^+]$ , a measure of the degree to which a solution is \_\_\_\_\_.

- A. Alkalinity
- B. Acidic or alkaline
- C. Hydrogen ion ( $H^+$ )
- D. None of the above

10. \_\_\_\_\_ is a substance that can give up a hydrogen ion ( $H^+$ ); a base is a substance that can accept  $H^+$ .

- A. Acid
- B. Base
- C. Acidic or alkaline
- D. None of the above

11. The more acidic a solution the greater the hydrogen ion concentration and the lower the pH; a pH of 7.0 indicates neutrality, a pH of less than 7 indicates acidity, and a pH of more than 7 indicates \_\_\_\_\_.

- A. Acid
- B. Base
- C. Alkalinity
- D. None of the above

## Turbidity Introduction

12. Generally, higher turbidity levels require higher coagulant dosages. However, seldom is the relationship between turbidity level and \_\_\_\_\_ linear.

- A. Coagulant dosage
- B. Total Dissolved Solids (TDS)
- C. Temperature
- D. None of the above

13. Usually, the extra coagulant required is relatively small when turbidities are much higher than normal due to higher collision probabilities of the \_\_\_\_\_ during high turbidities.

- A. Turbidity
- B. Colloids
- C. Total Dissolved Solids (TDS)
- D. None of the above

14. Low \_\_\_\_\_ waters can be very difficult to coagulate due to the difficulty in inducing collision between the colloids.

- A. Turbidity
- B. Colloids
- C. Total Dissolved Solids (TDS)
- D. None of the above

15. \_\_\_\_\_ may be existing in a water supply due to pollution, and these colloids can be difficult to remove in the coagulation process. In this situation, higher coagulant dosages are generally required.
- A. Turbidity                      C. Total Dissolved Solids (TDS)  
B. Organic colloids      D. None of the above

### **Turbidity MCL**

16. An MCL for turbidity established by the EPA because \_\_\_\_\_ interferes with disinfection. This characteristic of water changes the most rapidly after a heavy rainfall.
- A. Conductivity              C. Temperature  
B. Turbidity                  D. None of the above

17. The temperature variation of a sample, a scratched or unclean sample tube in the nephelometer and selecting an incorrect wavelength of a light path may be conditions caused by an inaccurate \_\_\_\_\_ measurement.
- A. Conductivity              C. Temperature  
B. Turbidity                  D. None of the above

### **Dissolved Oxygen**

18. The level of dissolved oxygen in natural waters is often a direct indication of quality, since aquatic plants produce oxygen, while microorganisms generally consume it as they feed on \_\_\_\_\_.
- A. Pollutants                  C. E. coli bacteria  
B. Organic matter      D. None of the above

19. Thermal stratification is possible as water becomes less dense when heated; meaning water weighs less per unit volume. Therefore, warmer water will be lighter and colder water will be heavier. Due to this, there will always be a level of “self-induced” \_\_\_\_\_ in a water storage.
- A. Saturation level(s)      C. Permanent hardness  
B. Thermal stratification      D. None of the above

### **Objections to Hard Water Scale Formation**

20. Hard water forms scale, usually \_\_\_\_\_, which causes a variety of problems. Left to dry on the surface of glassware and plumbing fixtures, including showers doors, faucets, and sink tops; hard water leaves unsightly white scale known as water spots.
- A. Magnesium carbonate      C. Calcite  
B. Calcium carbonate      D. None of the above

### **Secondary Standard**

21. TDS is most often measured in parts per million (ppm) or milligrams per liter of water (mg/L). The normal TDS level ranges from \_\_\_\_\_.
- A. 50 ppm to 1,000 ppm      C. 50 ppm to 100 ppm  
B. 5 ppm to 10 ppm      D. None of the above

### Langelier Saturation Index

22. The Langelier Saturation index (LSI) is an evenness scale derived from the theoretical concept of saturation and provides an indicator of the degree of saturation of water with respect to calcium carbonate. It can be shown that the Langelier saturation index (LSI) approximates the base 10 logarithm of the \_\_\_\_\_ saturation level.

- A. Magnesium carbonate
- B. Calcium sulfite
- C. Calcite
- D. None of the above

23. The Langelier saturation level approaches the concept of saturation using pH as a main variable. The LSI can be interpreted as the pH change required to bring water to \_\_\_\_\_.

- A. Saturation level(s)
- B. Stratification
- C. Equilibrium
- D. None of the above

### More on the Stage 2 DBP Rule

24. Which of the following rules focuses on public health protection by limiting exposure to DBPs, specifically total trihalomethanes and five haloacetic acids, that form in water through disinfectants used to control microbial pathogens?

- A. Stage 2 DBP rule
- B. Stage 1 DBPR
- C. Long Term 5 Enhanced Surface Water Treatment Rule
- D. None of the above

### Are THMs and HAAs the only disinfection byproducts?

25. The presence of TTHM and HAA5 is representative of the occurrence of many other chlorination DBPs; thus, an increase of TTHM and HAA5 generally indicate an increase of DBPs from chlorination.

- A. True
- B. False

### Disinfection Byproduct Research and Regulations Summary

26. \_\_\_\_\_ is unquestionably the most important step in the treatment of water for drinking water supplies.

- A. DBP(s)
- B. Turbidity (particle)
- C. Disinfection
- D. None of the above

### Controlling Disinfection Byproducts

27. Treatment techniques are available that provide water suppliers the opportunity to maximize potable water safety and quality while minimizing the risk of \_\_\_\_\_.

- A. DBP risks
- B. Turbidity (particle)
- C. Disinfectants and DBPs
- D. None of the above

28. Generally, the best approach to reduce \_\_\_\_\_ is to remove natural organic matter precursors prior to disinfection.

- A. DBP(s)
- B. Turbidity (particle)
- C. DBP formation
- D. None of the above

### Absorption

29. Activated carbon can be used to absorb \_\_\_\_\_ that react with disinfectants to form byproducts.

- A. Inorganic coagulants
- B. All contaminants
- C. Soluble organics
- D. None of the above

### Membrane Technology

30. Membranes, used historically to desalinate brackish waters, have also demonstrated excellent removal of \_\_\_\_\_.

- A. THMs and HAAs
- B. Optimization of pH
- C. Natural organic matter
- D. None of the above

31. EPA predicted that most water systems will be able to achieve compliance with new DBP regulations through the use of one or more of these relatively low cost methods (EPA, 1998). Water system managers may also consider switching from chlorine to alternative disinfectants to reduce formation of \_\_\_\_\_.

- A. THMs and HAAs
- B. Optimization of pH
- C. Natural organic matter
- D. None of the above

### Organisms Descriptors and Meanings

32. Photo means...

- A. Feed or nourish
- B. Other (Organic carbon)
- C. Light
- D. None of the above

33. Aerobic means...

- A. Without air
- B. With air
- C. Self (Inorganic carbon)
- D. None of the above

### Contaminants that may be present in sources of drinking water include:

34. Which of the following like salts and metals, which can be naturally occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining or farming?

- A. Radioactive contaminants
- B. Pesticides and herbicides
- C. Inorganic contaminants
- D. Microbial contaminants

35. Which of the following may come from a variety of sources such as agriculture, urban stormwater run-off, and residential uses?

- A. Radioactive contaminants
- B. Pesticides and herbicides
- C. Inorganic contaminants
- D. Microbial contaminants

36. Which of the following, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations and wildlife?

- A. Microbial contaminants
- B. Pesticides and herbicides
- C. Inorganic contaminants
- D. All of the above

### Dangerous Waterborne Microbes

37. Which of the following is a parasite that enters lakes and rivers through sewage and animal waste. It causes cryptosporidiosis, a mild gastrointestinal disease. The disease can be severe or fatal for people with severely weakened immune systems.

- A. Coliform Bacteria
- B. Cryptosporidium
- C. Giardia lamblia
- D. None of the above

38. Which of the following are not necessarily agents of disease, these may indicate the presence of disease-carrying organisms?

- A. Fecal coliform bacteria
- B. Cryptosporidium
- C. Shigella dysenteriae
- D. None of the above

39. Which of the following is a parasite that enters lakes and rivers through sewage and animal waste. It causes gastrointestinal illness (e.g. diarrhea, vomiting, and cramps)?  
 A. Coliform Bacteria C. Protozoa  
 B. Cryptosporidium D. None of the above
40. Which of the following is a species of the rod-shaped bacterial genus Shigella?  
 A. Fecal coliform bacteria C. Shigella dysenteriae  
 B. Cryptosporidium D. None of the above
41. Which of the following are common in the environment and are generally not harmful? However, the presence of these bacteria in drinking water are usually a result of a problem with the treatment system or the pipes which distribute water and indicates that the water may be contaminated with germs that can cause disease.  
 A. Coliform Bacteria C. Giardia lamblia  
 B. Cryptosporidium D. None of the above
42. Which of the following are bacteria whose presence indicates that the water may be contaminated with human or animal wastes? Microbes in these wastes can cause short-term effects, such as diarrhea, cramps, nausea, headaches, or other symptoms.  
 A. Fecal Coliform and E. coli C. Shigella dysenteriae  
 B. Cryptosporidium D. None of the above

### **Bacteriological Monitoring Introduction**

43. Which of the following are usually harmless, occur in high densities in their natural environment and are easily cultured in relatively simple bacteriological media?  
 A. Indicator bacteria C. Viruses  
 B. Amoebas D. None of the above
44. Indicators in common use today for routine monitoring of drinking water include total coliforms, fecal coliforms, and?  
 A. Cryptosporidium C. Escherichia coli (E. coli)  
 B. Protozoa D. None of the above
45. According to the text, the routine microbiological analysis of your water is for?  
 A. Contamination C. Coliform bacteria  
 B. Colloids D. None of the above

### **Bacteria Sampling**

46. Water samples for \_\_\_\_\_ must always be collected in a sterile container.  
 A. Amoebas C. Viruses  
 B. Bacteria tests D. None of the above

### **Methods**

47. The MMO-MUG test, a product marketed as \_\_\_\_\_, is the most common. The sample results will be reported by the laboratories as simply coliforms present or absent.  
 A. Colilert C. Total coliform analysis  
 B. Coliform D. None of the above

**The three (3) types of samples are:**

48. Samples collected following a coliform present routine sample. The number of repeat samples to be collected is based on the number of \_\_\_\_\_ samples you normally collect.

- A. Repeat      C. Routine
- B. Special     D. None of the above

49. A PWS has a second Level 1 Assessment within a rolling 12-month period.

- A. Trigger: Level 3 Assessment      C. All of the above
- B. Trigger: Level 2 Assessment      D. None of the above

50. A PWS on state-approved annual monitoring has a Level 1 Assessment trigger in 2 consecutive years.

- A. Trigger: Level 4 Assessment      C. All of the above
- B. Trigger: Level 2 Assessment      D. None of the above

**Positive or Coliform Present Results**

51. With a positive total coliform sample and after you have contacted an agency for assistance, you will be instructed as to the proper repeat sampling procedures and possible corrective measures for solving the problem. It is very important to initiate the \_\_\_\_\_ as the corrective measures will be based on those results.

- A. Perform routine procedures      C. Corrective measures
- B. Repeat sampling immediately     D. None of the above

**Heterotrophic Plate Count (Spread Plate Method)**

52. Which of the following provides a technique to quantify the bacteriological activity of a sample?

- A. Colonies      C. Heterotrophic Plate Count
- B. Agar          D. None of the above

**The following are acute violations:**

53. Which determines a violation of nitrate?

- A. Presence      C. MCLG
- B. MCL          D. None of the above

**Revised Total Coliform Rule (RTCR) Summary**

54. The water provider shall develop and follow a sample-siting plan that designates the PWS's collection schedule. This includes location of \_\_\_\_\_.

- A. Routine and repeat water samples      C. Microbial contamination
- B. Reduced monitoring                      D. Repeat water samples

55. The water provider shall collect \_\_\_\_\_ on a regular basis (monthly, quarterly, annually). Have samples tested for the presence of total coliforms by a state certified laboratory.

- A. Routine water samples      C. Microbial contamination
- B. Reduced monitoring        D. Repeat water samples

56. PWSs incur violations if they do not comply with the requirements of the RTCR. The violation types are essentially the same as under the TCR with few changes. The biggest change is no acute or monthly MCL violation for \_\_\_\_\_ only.

- A. CCR(s)
- B. PN
- C. Total coliform positive samples
- D. TC+ routine or repeat sample

57. Community water systems (CWSs) must use specific language in their CCRs when they must conduct an assessment or if they incur \_\_\_\_\_.

- A. CCR(s)
- B. PN
- C. An E. coli MCL violation
- D. TC+ routine or repeat sample

58. The water provider shall analyze all \_\_\_\_\_ that are total coliform positive (TC+) for E. coli.

- A. Routine or repeat water samples
- B. Reduced monitoring
- C. Microbial contamination
- D. None of the above

### Disinfection Key

59. The RTCR requires 99.99% or 4 log inactivation of \_\_\_\_\_.

- A. Enteric viruses
- B. Crypto
- C. Giardia lamblia cysts
- D. None of the above

60. The RTCR requires 99% or 2 log inactivation of \_\_\_\_\_.

- A. Enteric viruses
- B. Crypto
- C. Giardia lamblia cysts
- D. None of the above

### Waterborne Pathogen Section - Introduction

#### Pathogen Section

61. Most pathogens are generally associated with diseases that \_\_\_\_\_ and affect people in a relatively short amount of time, generally a few days to two weeks.

- A. Cause intestinal illness
- B. Are mild in nature
- C. Will cause fatalities
- D. None of the above

#### How Diseases are Transmitted.

62. Waterborne pathogens are primarily spread by the?

- A. Fecal-oral, or feces-to-mouth route
- B. Dermal to fecal route
- C. Oral to fecal route
- D. None of the above

#### Protozoan Caused Diseases

63. Which of the following bugs is larger than bacteria and viruses but still microscopic; they invade and inhabit the gastrointestinal tract?

- A. Hepatitis A
- B. E.coli
- C. Protozoan pathogens
- D. None of the above

64. Some of the parasites enter the environment in a dormant form, with a protective cell wall, called a?

- A. Lamblia
- B. Shell
- C. Cyst
- D. None of the above

### **Giardia lamblia**

65. Which of the following bugs has been responsible for more community-wide outbreaks of disease in the U.S. than any other, and drug treatment are not 100% effective?

- A. Giardia lamblia
- B. Cryptosporidiosis
- C. Giardiasis
- D. None of the above

66. All of these diseases, with the exception of \_\_\_\_\_, have one symptom in common: diarrhea. They also have the same mode of transmission, fecal-oral, whether through person-to-person or animal-to-person contact.

- A. HIV infection
- B. Giardiasis
- C. Hepatitis A
- D. None of the above

### **Primary Waterborne Diseases Section**

67. Humans are the reservoir for the Salmonella typhi pathogen, which causes diarrheal illness, and also known as?

- A. Campylobacter
- B. Shigella dysenteriae
- C. Typhoid fever
- D. None of the above

68. Legionnaire's disease, which causes a severe pneumonia, and the second, \_\_\_\_\_, which is a non-pneumonia illness; it's typically an influenza-like illness, and it's less severe.

- A. Pontiac fever
- B. Yellow fever
- C. Beaver fever
- D. None of the above

69. Humans are the reservoir for the Norovirus. Prevention strategies for this pathogen include?

- A. Internal protection
- B. Source protection
- C. Containment protection
- D. None of the above

70. Giardia prevention strategies for this pathogen include \_\_\_\_\_; filtration, coagulation, and halogenation of drinking water.

- A. Internal protection
- B. Source protection
- C. Containment protection
- D. None of the above

### **Chain of Custody Procedures**

71. If both parties involved in the transfer must sign, date and note the time on the chain of custody record, this is known as?

- A. TC Plan
- B. Sample siting plan
- C. Samples transfer possession
- D. None of the above

72. The recipient will then attach the \_\_\_\_\_ showing the transfer dates and times to the custody sheets. If the samples are split and sent to more than one laboratory, prepare a separate chain of custody record for each sample.

- A. Shipping invoices
- B. Chain of custody release
- C. Sample siting plan
- D. None of the above

### Factors in Chlorine Disinfection: Concentration and Contact Time

73. Based on the work of several researchers, CXT values [ final free chlorine concentration (mg/L) multiplied by minimum contact time (minutes)], offer water operators guidance in computing an effective combination of chlorine concentration and \_\_\_\_\_ required to achieve disinfection of water at a given temperature.

- A. Chlorine concentration
- B. Chlorine contact time
- C. Higher strength chlorine solutions
- D. None of the above

74. The CXT formula demonstrates that if an operator chooses to decrease the chlorine concentration, the required \_\_\_\_\_ must be lengthened.

- A. Chlorine concentration
- B. Temperature
- C. Contact time
- D. None of the above

75. As \_\_\_\_\_ are used, contact times may be reduced.

- A. Chlorine concentration
- B. Temperature
- C. Higher strength chlorine solutions
- D. None of the above

### Water Treatment Section - Preliminary Treatment Process

#### Preliminary Treatment

76. Weeds, leaves, and trash, if not removed, these will cause problems to the treatment plant's pumps and equipment, the best way to protect the plant is?

- A. Screening
- B. Super settling
- C. Change source
- D. None of the above

77. According to the text, wire mesh screens need maintenance and require?

- A. Manual cleaning
- B. PM cleaning
- C. No cleaning
- D. None of the above

#### Pre-Sedimentation

78. Sand and grit will damage plant equipment and pipes, so it must be removed with either rectangular or round shaped basin are called?

- A. Filtration basin(s)
- B. Coagulation basin(s)
- C. Sedimentation basin(s)
- D. None of the above

79. Which of the following treatment terms is used after the flocculation process?

- A. Filtration basin(s)
- B. Coagulation basin(s)
- C. Sedimentation basin(s)
- D. None of the above

80. Scrapers on the bottom move the settled sludge to one or more hoppers at the influent end of the tank; it may have a \_\_\_\_\_ or traveling bridge used to collect the sludge.

- A. Screw conveyor
- B. Conveyor belts
- C. Manual skimmer
- D. None of the above

#### Flights and Chains

81. Flights and chains remove the scum from the \_\_\_\_\_ of the basin.

- A. Scum box
- B. Surface
- C. Armature
- D. None of the above

(S) Means the answer can be plural or singular in nature

### Circular Clarifiers

82. Which of the following systems use graded silica sand filter media?  
A. Traditional filter systems                      C. Chemical pretreatment  
B. Reconditioning cycle                              D. None of the above
83. Which of the following processes uses alum and cationic polymer to neutralize the charge of colloidal particles?  
A. Filtration    C. Flocculation  
B. Reconditioning                                      D. None of the above
84. Which of the following compounds combine with alkalinity in the raw water to form a white precipitate that neutralizes suspended particles' electrical charge?  
A. Activated sodium                                      C. Alum  
B. PAC    D. None of the above
85. Which of the following terms may increase filtered water clarity, measured in NTU, by 90% compared with filtration alone?  
A. Chemical pretreatment                              C. Fast rinse  
B. Reconditioning cycle                                      D. None of the above
86. Water treatment systems use settling tanks unit to allow for \_\_\_\_\_.  
A. Gravity    C. Settling time  
B. Particle(s)    D. Sedimentation and settling
87. The main aim of tube settlers is to minimize the \_\_\_\_\_ that a small floc particle must settle before agglomerating into larger particles.  
A. Gravity    C. Settling time  
B. Vertical distance    D. Solids

### Conventional Water Treatment Process Introduction

88. \_\_\_\_\_ along with pre-chlorination for removal of dissolved iron when present with small amounts relative of manganese  
A. Disinfection    C. Pre-treatment  
B. Coagulation    D. Aeration
89. \_\_\_\_\_ to remove particles from water either by passage through a sand bed that can be washed and reused or by passage through a purpose- designed filter that is washable.  
A. Disinfection    C. Pre-treatment  
B. Coagulation    D. Filtration
90. \_\_\_\_\_ for killing bacteria viruses and other pathogens.  
A. Disinfection    C. Pre-treatment  
B. Coagulation    D. Aeration along with pre-chlorination
91. \_\_\_\_\_ or slow-sand filtration  
A. Disinfection    C. Pre-treatment  
B. Coagulation    D. Coagulation or flocculation

## Coagulants – Alum and Ferric

### Aluminum Sulfate (Alum)

92. Once in water, alum can react with hydroxides, carbonates, bicarbonates, and other anions to form \_\_\_\_\_.

- A. pH
- B. Alkalinity
- C. Large, positively charged molecules
- D. None of the above

93. Carbon dioxide and sulfate are generally byproducts of these reactions. During the reactions, alum acts as \_\_\_\_\_ to reduce the pH and alkalinity of the water supply. It is important that sufficient alkalinity be present in the water supply for the various reactions to occur.

- A. Inorganic coagulant(s)
- B. An acid
- C. Byproducts of these reactions
- D. None of the above

94. The aluminum ions become soluble rather than insoluble and do not participate in the hydration and \_\_\_\_\_ necessary to make the alum effective as a coagulant. In these instances, the plant may experience higher than normal filtered water turbidities, and much of the aluminum will pass through the filters.

- A. Post filtration alum coagulation
- B. Olation reaction(s)
- C. Byproducts of these reactions
- D. None of the above

95. When the pH level of the water is above 7.8 after the addition of the alum, the aluminum ions again become soluble, and the efficiency of coagulation is decreased. Under these conditions, aluminum ions again penetrate the filters, and \_\_\_\_\_ can occur in the clear well and in the distribution system in some cases.

- A. Post filtration alum coagulation
- B. Olation reaction(s)
- C. Byproducts of these reactions
- D. None of the above

### Ferric Chloride (Ferric)

96. Like ferric sulfate, ferric chloride exhibits a wide \_\_\_\_\_ range for coagulation, and the ferric ion does not easily become soluble.

- A. pH
- B. Alkalinity
- C. Olation
- D. None of the above

97. As a result, many plants are replacing alum with ferric chloride to eliminate the penetration of aluminum ions through the plant filters. Ferric chloride also reacts as an acid in water to reduce \_\_\_\_\_.

- A. pH
- B. Alkalinity
- C. Olation
- D. None of the above

98. \_\_\_\_\_ are available, such as potash alum, ammonia alum, ferrous sulfate (copperas), and chlorinated copperas.

- A. Other inorganic coagulants
- B. Olation reaction(s)
- C. Byproducts of these reactions
- D. None of the above

99. Typical dosages of the inorganic coagulants range from 50 pounds per million gallons of water treated under ideal conditions to as high as 800 to 1000 pounds per million gallons of water treated under \_\_\_\_\_ conditions.

- A. Worst case
- B. Decreased
- C. Increased
- D. None of the above

## Factors Influencing Coagulation

### Effects of pH

100. The pH range in which a coagulation process occurs may be the single most important factor in \_\_\_\_\_ coagulation. The vast majority of coagulation problems are related to improper pH levels.

- A. Improper
- B. Optimum
- C. Proper
- D. None of the above

101. Whenever possible, coagulation should be conducted in \_\_\_\_\_. When this is not done, lower coagulation efficiency results, generally resulting in a waste of chemicals and a lowered water quality.

- A. The optimum pH zone
- B. The coagulation process
- C. Collision between the colloids
- D. None of the above

102. Each of the inorganic salt coagulants has its own characteristic \_\_\_\_\_ pH range.

- A. Improper
- B. Optimum
- C. Little or no effect
- D. None of the above

103. In many plants, it is necessary to adjust the pH level in the coagulation process. In most cases, this involves the addition of lime, caustic soda, or soda ash to maintain a minimum pH level. In some cases, however, acids may be necessary to raise or lower the pH level to an \_\_\_\_\_ range.

- A. Improper
- B. Optimum
- C. Little or no effect
- D. None of the above

104. In some water plants, the acidic reactions of the inorganic salts are taken advantage of when the raw water pH levels are \_\_\_\_\_. In these instances, overfeed of the coagulant is intentionally induced in order for the coagulation process to occur in the optimum range.

- A. Improper
- B. Optimum
- C. Higher than desired
- D. None of the above

### Effects of Salts

105. Since no natural waters are completely pure, each will have various levels of cations and anions such as calcium, sodium, magnesium, iron, manganese, sulfate, chloride, phosphate, and others. Some of these ions may affect the efficiency of \_\_\_\_\_.

- A. All chemical reactions
- B. The coagulation process
- C. Collision between the colloids
- D. None of the above

106. Trivalent cations do not have an adverse effect on the process in most instances. In fact, significant concentrations of naturally occurring iron in a water supply has resulted in the ability to feed \_\_\_\_\_ dosages of inorganic salt coagulants.

- A. Improper
- B. Optimum
- C. Lower than normal
- D. None of the above

### Nature of Turbidity

107. Generally, higher turbidity levels require higher coagulant dosages. However, seldom is the relationship between turbidity level and coagulant dosage linear. Usually, the additional coagulant required is \_\_\_\_\_ when turbidities are much higher than normal due to higher collision probabilities of the colloids during high turbidities.

- A. Improper
- B. Optimum
- C. Relatively small
- D. None of the above

108. Conversely, low turbidity waters can be very difficult to coagulate due to the difficulty in inducing \_\_\_\_\_.

- A. All chemical reactions
- B. The coagulation process
- C. Collision between the colloids
- D. None of the above

109. In the above instance, \_\_\_\_\_ formation is poor, and much of the turbidity is carried directly to the filters.

- A. Poor
- B. Average
- C. Good
- D. None of the above

110. Organic colloids may be present in a water supply due to pollution, and these colloids can be difficult to remove in the coagulation process. In this situation, \_\_\_\_\_ coagulant dosages are generally required.

- A. Improper
- B. Higher
- C. Slowly
- D. None of the above

### Water Temperature

111. Cold water temperatures can cause two factors which add to the difficulty of the coagulation process. As water temperatures approach freezing, almost all chemical reactions occur more \_\_\_\_\_.

- A. Improper
- B. Higher
- C. Slowly
- D. None of the above

### Corrosion Control Introduction

112. Corrosion is the deterioration of a substance by chemical action. Lead, cadmium, zinc, copper and iron might be found in water when metals in water distribution systems corrode. Drinking water contaminated with certain metals (such as \_\_\_\_\_) can harm human health.

- A. Lead
- B. Lead and copper
- C. Lead and cadmium
- D. None of the above

113. The EPA has banned the use of lead solders, fluxes and pipes in the installation or repair of any public water system. In the past, solder used in plumbing has been \_\_\_\_\_.

- A. 60% lead and 40% tin
- B. 50% tin and 50% lead
- C. 50% copper and 50% lead
- D. None of the above

114. Using lead-free solders, such as \_\_\_\_\_ is a key factor in lead corrosion control.

- A. 20% lead and 80% tin
- B. Silver-tin and antimony-tin
- C. Lead and cadmium
- D. None of the above

## Coagulation and Flocculation Summary

### Rapid Sand Filtration

115. Which term is the most prevalent form of water treatment technology in use today?

- A. Conventional technology
- B. Sedimentation process
- C. Rapid Sand filtration
- D. None of the above

116. Rapid Sand filtration process employs a combination of \_\_\_\_\_ in order to achieve maximum effectiveness.

- A. Filtration
- B. Sedimentation process
- C. Physical and chemical processes
- D. None of the above

### Coagulation

117. The alum and the water are mixed rapidly by the?

- A. Cationic polymers
- B. Flash mixer
- C. Shaker
- D. None of the above

118. What is the process of joining together particles in water to help remove organic matter called?

- A. Cationic binding
- B. Coagulation
- C. Flocculation
- D. None of the above

119. Fine particles must be coagulated, or "stuck together" to form larger particles that can be filtered, this is achieved through the use of?

- A. Sedimentation chemicals
- B. Coagulant chemicals
- C. Flocculation chemicals
- D. None of the above

120. Which of the following terms are required since colloidal particles by themselves have the tendency to stay suspended in water and not settle out?

- A. Sedimentation chemicals
- B. Coagulant chemicals
- C. Flocculation chemicals
- D. None of the above

121. Which of the following terms are so small, their charge per volume is significant?

- A. Aluminum Sulfate molecules
- B. Coagulant chemicals
- C. Colloidal particles
- D. None of the above

### Flocculation

122. Flocculation is the process where the suspended particles can collide, \_\_\_\_\_, and form heavier particles called "floc".

- A. Equalization
- B. Agglomerate
- C. Destabilized or coagulated particles
- D. None of the above

123. Gentle \_\_\_\_\_ and appropriate detention times (the length of time water remains in the basin) help facilitate the flocculation process.

- A. Equalizing
- B. Agitation of the water
- C. Settling
- D. None of the above

124. Which of the following happens in the water when bacteria and other microorganisms are caught in the floc structure?

- A. Equalize the basin
- B. Floc particles mix
- C. Agitate the water
- D. None of the above

### Pre-Sedimentation

125. Contingent on the quality of the source water, some plants have pre-sedimentation, which allows larger \_\_\_\_\_ in a reservoir or lake reducing solid removal loads.

- A. Equalization of the basin
- B. Particles time to settle
- C. Floc particles mix
- D. None of the above

### Water Filtration Key Terms

#### Declining Rate Filters

126. The filter flow rate will vary with?

- A. Head loss
- B. Uniform media
- C. Effluent control
- D. None of the above

127. Declining Rate Filters system often requires \_\_\_\_\_ to provide adequate media submergence.

- A. Head loss
- B. Uniform media
- C. Effluent control structure
- D. None of the above

### Disinfection

128. Chlorine is added again after filtration for?

- A. Residual
- B. Contact time
- C. Post-disinfection
- D. None of the above

### pH

129. According to the text, which of the following has a pH between 6.0 and 8.5?

- A. Acids
- B. Disinfectants
- C. Natural water
- D. None of the above

### Taste and Odor Control

130. Which of the following is occasionally added for taste and odor control?

- A. Turbidity powder
- B. Fluoride
- C. Powdered activated carbon (PAC)
- D. None of the above

### Short-Circuiting

131. Short-Circuiting is usually undesirable, since it may result in shorter contact, reaction, or settling times in comparison with the?

- A. Presumed detention times
- B. Sedimentation/clarification process
- C. Modification of the conventional process
- D. None of the above

### Tube Settlers

132. Tube settlers are a modification of the conventional process contains many metal "tubes" that are normally placed in?

- A. Flocculation basin
- B. Sedimentation basin or clarifier
- C. An up-flow clarifier
- D. None of the above

133. The slope of the tube settlers facilitates gravity settling of the solids to the bottom of the basin, where they can be?

- A. Adjusted for detention times
- B. Modified
- C. Collected and removed
- D. None of the above

### **Filtration Overview**

134. According to the text, the filter is periodically cleaned by a reversal of flow and the \_\_\_\_\_ into a drain.

- A. Activated carbon filters
- B. Anthracite coal
- C. Rapid-sand filters
- D. None of the above

### **EPA Filter Backwash Rule- Introduction**

#### **Turbidity**

135. Which of the following must comply with specific combined filter effluent turbidity requirements?

- A. Watershed
- B. Disinfection profile
- C. Conventional and Direct filtration systems
- D. None of the above

#### **Disinfection Benchmarking**

136. Public water systems will be required to develop a \_\_\_\_\_ unless they perform applicability monitoring which demonstrates their disinfection byproduct levels are less than 80% of the maximum contaminant levels.

- A. Disinfection profile
- B. Direct filtration system
- C. Disinfection benchmark
- D. None of the above

137. According to the text, if a system considers making a significant change to their disinfection practice, they must develop a(n) \_\_\_\_\_ and receive State approval for implementing the change.

- A. Disinfection profile
- B. Direct filtration systems
- C. Disinfection benchmark
- D. None of the above

138. Which of the following that practice direct recycle, employ 20 or fewer filters to meet production requirements during a selected month, and recycle spent filter backwash water, thickener supernatant, and/or liquids from the dewatering process within the treatment process must perform a one month, one-time recycle self-assessment?

- A. Recycle systems
- B. Conventional systems
- C. Direct filtration systems
- D. None of the above

139. Which of the following will be required to return spent filter backwash water, thickener supernatant, and liquids from the dewatering process prior to the point of primary coagulant addition unless the State specifies an alternative location?

- A. Recycle systems
- B. Conventional systems
- C. Direct filtration systems
- D. None of the above

140. Which of the following recycling to the treatment process must provide detailed recycle treatment information to the State, which may require that modifications to the recycle practice be made?

- A. Recycle systems
- B. Conventional systems
- C. Direct filtration systems
- D. None of the above

#### **Filtration Process- Detailed**

141. Removal of \_\_\_\_\_ plays an important role in the natural treatment of groundwater as it percolates through the soil.

- A. Suspended solids by filtration
- B. Serious problems in filter operation
- C. Coagulation and flocculation processes
- D. None of the above

142. According to the text, since surface water sources are subject to run-off and do not undergo natural filtration, it must be filtered to?

- A. Remove particles and impurities
- B. Increase chlorination
- C. Increase suspended particles
- D. None of the above

143. Which of the following traps suspended material between the grains of filter media?

- A. Bubble diffuser
- B. Filtration process - Compared to a sieve or microstrainer
- C. Retention Basins - Suspended particles can easily pass
- D. None of the above

144. Which of the following will easily pass through the spaces between the grains of the filter media, this making straining the least important process in filtration?

- A. Heavy particles and impurities
- B. Sludge
- C. Suspended particles
- D. None of the above

145. Adsorption is the process of particles sticking onto the surface of the individual filter grains or onto the previously deposited materials. The forces that attract and hold the particles to the grains are the same as those that work in \_\_\_\_\_.

- A. Coagulation and flocculation
- B. Filter operation
- C. Chlorination
- D. None of the above

146. Which of the following may occur in the filter bed will happen especially if coagulation and flocculation of the water before filtration was not properly controlled?

- A. Coagulation and flocculation
- B. Filter operation
- C. Flocculation
- D. None of the above

#### **Direct Filtration Plant vs. Conventional Plant**

147. The primary difference between Direct Filtration Plant vs. Conventional Plant is that the \_\_\_\_\_ or step is omitted from the Direct Filtration plant.

- A. Sedimentation process
- B. Reconditioning cycle
- C. Fast rinse
- D. None of the above

#### **Types of Filters**

148. What is the term for the mass of growing material that collects on the surface of the filter?

- A. Schmutzdecke
- B. Zoological growth
- C. Mud balls
- D. None of the above

149. Most water filters are classified by filtration rate, type of \_\_\_\_\_, or type of operation.

- A. Schmutzdecke
- B. Backwash capabilities
- C. Filter media
- D. None of the above

(S) Means the answer can be plural or singular in nature

### Rapid Sand Filters

150. Rapid sand filters can accommodate filter rates 40 times more than?

- A. Fixed film
- B. Slow sand filters
- C. Mixed media
- D. None of the above

### Filter Sand

151. Which of the following will contain 24-30 inches of sand, but some newer filters are deeper?

- A. Rapid sand filters
- B. Slow rate filters
- C. Sedimentation basins
- D. None of the above

152. The coarser sand in the \_\_\_\_\_ has larger voids that do not fill as easily.

- A. Rapid filters
- B. Backwash trough
- C. Sedimentation basin
- D. None of the above

### False floor

153. The false floor design of a \_\_\_\_\_ is used together with a porous plate design or with screens that retain the sand when there is no undergravel layer.

- A. Backwash system
- B. Leopold system
- C. Filter underdrain
- D. None of the above

### Pressure Sand Filters

154. Which of the following terms or methods cracking of the filter bed can occur quite easily, allowing the iron and manganese particles to go straight through the filter?

- A. Slow sand/RO
- B. Gravity filters
- C. Pressure filters
- D. None of the above

155. Which of the following filtration types is contained under pressure in a steel tank?

- A. Slow sand/RO
- B. Gravity filters
- C. Pressure sand filter
- D. None of the above

156. In which of the following filtration types is the media usually sand or a combination of media?

- A. Slow sand/RO
- B. Gravity filters
- C. Fast sand
- D. None of the above

157. Which of the following filter types has a major disadvantage in that the backwash cannot be observed?

- A. Slow sand/RO
- B. Gravity filters
- C. Pressure filters
- D. None of the above

158. Filtration operation is divided into three steps: filtering, backwashing, and?

- A. Filter run
- B. Filtering to waste
- C. Return to waste
- D. None of the above

159. Which of the following is a low-pressure membrane filtration process that removes suspended solids and colloids generally larger than 0.1-micron diameter?

- A. Nanofiltration
- B. Microfiltration
- C. Semi-permeable
- D. None of the above

160. Which of the following is a relatively recent membrane process used most often with low total dissolved solids water such as surface water and fresh groundwater?

- A. Nanofiltration
- B. Microfiltration
- C. Semi-permeable
- D. None of the above

### Declining Rate

161. According to the text, which of the following methods of control is used where the largest head loss occurs in the filtration process?

- A. Declining Rate
- B. Gravity filters
- C. Fast sand
- D. None of the above

162. The rate through the declining filter is much greater in the beginning of a filter run than at the end when the?

- A. Filter run
- B. Filter is dirty
- C. Head loss is low
- D. None of the above

### Loss of Head Indicator

163. Which of the following is required to force the water through the filter?

- A. Filter run
- B. Force
- C. Head loss
- D. None of the above

### Filtration Process

164. The filter eventually fills with suspended material, usually after 15 to 30 hours, it will need to be \_\_\_\_\_ to clean the media.

- A. Bumped
- B. Jetted
- C. Backwashed
- D. None of the above

### Back Washing

165. Which of the following if it is too high will cause media to be washed from the filter into the troughs and out of the filter.

- A. Media
- B. Floc(s)
- C. Backwash rate
- D. None of the above

### Backwashing Process

166. The backwash valve is opened, allowing backwash water to start flowing into the filter and start carrying \_\_\_\_\_ away from the filter.

- A. Headloss
- B. Crust on the filter
- C. Suspended material
- D. None of the above

### Disposal of Filter Backwash Water

167. The supernatant is then pumped back to the head of the treatment plant at a rate not exceeding ten percent of the?

- A. Daily flow
- B. Backwash water
- C. Raw water flow entering the plant
- D. None of the above

### Filter Aids

168. Which of the following terms expresses that the polymer strengthens the bonds and prevents the shearing forces in the filter from breaking the floc apart when used?

- A. Filter media
- B. Lime
- C. Filter aid
- D. None of the above

### Filter Operating Problems

169. According to the text, there are three major types of filter problems. They can be caused by chemical treatment before the filter, \_\_\_\_\_, and backwashing of filters.

- A. Filter aid
- B. Control of filter flow rate
- C. Coagulation and flocculation stages
- D. None of the above

### Chemical Treatment before the Filter

170. Which of the following terms of the water treatment must be monitored continuously?

- A. Filter aid
- B. Backwash storage basin
- C. Coagulation and flocculation stages
- D. None of the above

### Advanced Water Treatment Section

171. Water contains \_\_\_\_\_ of which impart a quality known as hardness?

- A. TDS
- B. Conductivity
- C. Various amounts of dissolved minerals
- D. None of the above

### Occurrence of Hard Water

172. Which of the following is caused by soluble, divalent, metallic cations, (positive ions having valence of 2)?

- A. Hard water
- B. Permanent hardness
- C. Carbonate hardness
- D. None of the above

### Types of Hardness

173. Hardness can be categorized by either of two methods: calcium versus magnesium hardness and?

- A. Carbonate hardness
- B. Temporary hardness
- C. Carbonate versus non-carbonate hardness
- D. None of the above

### Carbonate-Noncarbonate Distinction

174. According to the text, the carbonate-noncarbonate distinction, is based on hardness from either the bicarbonate salts of calcium or the \_\_\_\_\_ involved in causing water hardness.

- A.  $\text{CaCO}_3$
- B. Water hardness
- C. Normal salts of calcium and magnesium
- D. None of the above

### Membrane Filtration Processes

175. Which of the following enables some water systems having contaminated water sources to meet new, more stringent regulations?

- A. Membrane technology
- B. Macromolecule(s)
- C. Conventional thermal separation process(es)
- D. None of the above

### Nanofiltration

176. Nanofiltration (NF) process has been used primarily for water softening and reduction of?

- A. Process liquid
- B. Bacterial and protozoan life
- C. Total dissolved solids (TDS)
- D. None of the above

## Reverse Osmosis

177. RO membranes have very low MWC pore size that can reject ions at very high rates, including?

- A. Process liquid
- B. Chloride and sodium
- C. Bacterial and protozoan life
- D. None of the above

## Water Laboratory Analysis Section

### pH Testing Section

178. Mathematically, pH is the negative logarithm of the activity of the (solvated) hydronium ion, more often expressed as the measure of the?

- A. Electron concentration
- B. Alkalinity concentration
- C. Hydronium ion concentration
- D. None of the above

179. Alkalinity is the name given to the quantitative capacity of an aqueous solution to neutralize an?

- A. Acid
- B. Base
- C. Bond formation
- D. None of the above

10. Sodium hydroxide, NaOH, is an example of a?

- A. Weak base
- B. Strong base
- C. Strong acid
- D. None of the above

### Disinfection Section

#### Chlorine's Appearance and Odor

181. Chlorine is a greenish-yellow gas it will condense to an amber liquid at approximately \_\_\_\_\_ F or at high pressures.

- A. -29.2 degrees
- B. - 100 degrees
- C. 29 degrees
- D. None of the above

182. Prolonged exposures to chlorine gas may result in?

- A. Moisture, steam, and water
- B. Odor thresholds
- C. Olfactory fatigue
- D. None of the above

### Chlorine Gas

#### Pathophysiology

183. The odor threshold for chlorine gas is approximately?

- A. 0.3-0.5 parts per million (ppm)
- B. 3 parts per million (ppm)
- C. 30-500 parts per million (ppm)
- D. None of the Above

### Chlorination Chemistry

184. According to the text, pH and temperature affect the ratio of hypochlorous acid to hypochlorite ions. As the temperature is decreased, the \_\_\_\_\_ increases.

- A. Reduction Ratio
- B. Ratio of hypochlorous acid
- C. "CT" disinfection concept
- D. None of the above

185. Although the ratio of \_\_\_\_\_ is greater at lower temperatures, pathogenic organisms are actually harder to kill.

- A. Hypochlorous acid
- B. The amount of chlorine
- C. Total chlorine
- D. None of the above

186. If all other things were equal, \_\_\_\_\_ and a lower pH are more conducive to chlorine disinfection.

- A. Lower pH
- B. Hypochlorous acid
- C. Higher water temperatures
- D. None of the above

### Chlorine DDBP

187. These term means that chlorine is present as Cl, HOCl, and OCl<sup>-</sup> is called \_\_\_\_\_, and that which is bound but still effective is \_\_\_\_\_.

- A. Free available chlorine and Total
- B. Free and Residual
- C. Free available chlorine and Combined Chlorine
- D. None of the above

188. Chloramines are formed by reactions with?

- A. Acid and Cl<sub>2</sub>
- B. Ammonia and Cl<sub>2</sub>
- C. Folic Acid and Cl<sub>2</sub>
- D. None of the above

### Types of Residual

189. Which of the following is all chlorine that is available for disinfection?

- A. Chlorine residual
- B. Chlorine demand
- C. Total chlorine
- D. None of the Above

### Chlorine Exposure Limits

190. What is OSHA's PEL?

- A. 10 PPM
- B. 1 PPM
- C. 1,000 PPM
- D. None of the above

191. Liquid chlorine is about \_\_\_\_\_ times heavier than water

- A. 1.5
- B. 10
- C. 2.5
- D. None of the above

192. Gaseous chlorine is about \_\_\_\_\_ times heavier than air.

- A. 1.5
- B. 10
- C. 2.5
- D. None of the above

### Alternate Disinfectants - Chloramine

193. It is recommended that Chloramine be used in conjunction with a stronger disinfectant. It is best utilized as a?

- A. Chloramine
- B. T10 value disinfectant
- C. Stable distribution system disinfectant
- D. None of the above

194. In the production of \_\_\_\_\_, the ammonia residuals in the finished water, when fed in excess of stoichiometric amount needed, should be limited to inhibit growth of nitrifying bacteria.

- A. Dry sodium chlorite
- B. Chloramines
- C. Ammonia residual(s)
- D. None of the above

### Chlorine Dioxide

195. Which term provides good Giardia and virus protection, but its use is limited by the restriction on the maximum residual of 0.5 mg/L ClO<sub>2</sub>/chlorite/chlorate allowed in finished water?

- A. Chlorinated byproducts
- B. Chlorine dioxide
- C. Ammonia residual(s)
- D. None of the above

196. If chlorine dioxide is being used as an oxidant, the preferred method of generation is to entrain this term or substance into a packed reaction chamber with a 25% aqueous solution of sodium chlorite (NaClO<sub>2</sub>).

- A. Chloramine
- B. Chlorine gas
- C. Chlorine dioxide
- D. None of the above

197. According to the text, which chemical is explosive and can cause fires in feed equipment if leaking solutions or spills are allowed to dry out?

- A. Dry sodium chlorite
- B. Chlorine dioxide
- C. Ammonia
- D. None of the above

198. Chlorine dioxide may be used for either taste or odor control or as a?

- A. Chloramine
- B. Pre-disinfectant
- C. Gas
- D. None of the above

### Ozone

199. When determining Ozone CT (contact time) values must be determined for the ozone basin alone; an accurate \_\_\_\_\_ must be obtained for the contact chamber, and residual levels.

- A. Residual
- B. T10 value
- C. Contact time
- D. None of the above

200. Ozone does not provide a system residual and should be used as a primary disinfectant only in conjunction with?

- A. Dry sodium chlorite
- B. Chlorine dioxide
- C. Free and/or combined chlorine
- D. None of the above

## When Finished with Your Assignment...

### REQUIRED DOCUMENTS

Please scan the **Registration Page, Answer Key, Proctoring reports, Survey and Driver's License** and email these documents to [info@TLCH2O.com](mailto:info@TLCH2O.com).

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